



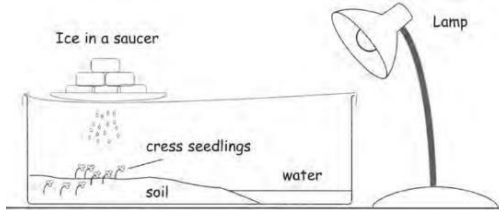
## Chemistry

### KS1 Working Scientifically:

1. asking simple questions and recognising that they can be answered in different ways
2. observing closely, using simple equipment
3. performing simple tests
4. identifying and classifying
5. using their observations and ideas to suggest answers to questions
6. gathering and recording data to help in answering questions

Year 1	Knowledge	Vocabulary	Skills
Autumn Term	<p><b><u>Everyday materials</u></b></p> <ul style="list-style-type: none"> <li>distinguish between an object and the material from which it is made</li> <li>identify and name a variety of everyday materials, including wood, plastic, glass, metal, water, and rock</li> <li>describe the simple physical properties of a variety of everyday materials</li> <li>compare and group together a variety of everyday materials on the basis of their simple physical properties</li> </ul>	Materials, objects, wood, plastic, metal, liquid, gas, solid, stretch, bend, waterproof, shiny	<ol style="list-style-type: none"> <li>1. <i>e.g. which material would be most suitable for a waterproof coat?</i></li> <li>4. <i>e.g. identify a range of classroom objects and classify based on the material they are made from</i></li> <li>5. <i>e.g. why is the table made from wood and not glass?</i></li> </ol>

LKS2 Working Scientifically:	
1.	asking relevant questions and using different types of scientific enquiries to answer them
2.	setting up simple practical enquiries, comparative and fair tests
3.	making systematic and careful observations and, where appropriate, taking accurate measurements using standard units, using a range of equipment, including thermometers and data loggers
4.	gathering, recording, classifying and presenting data in a variety of ways to help in answering questions
5.	recording findings using simple scientific language, drawings, labelled diagrams, keys, bar charts, and tables
6.	reporting on findings from enquiries, including oral and written explanations, displays or presentations of results and conclusions
7.	using results to draw simple conclusions, make predictions for new values, suggest improvements and raise further questions
8.	identifying differences, similarities or changes related to simple scientific ideas and processes
9.	using straightforward scientific evidence to answer questions or to support their findings

Year 4	Knowledge	Vocabulary	Skills
Summer term	<p><b>States of Matter</b></p> <ul style="list-style-type: none"> <li>compare and group materials together, according to whether they are solids, liquids or gases</li> <li>observe that some materials change state when they are heated or cooled, and measure or research the temperature at which this happens in degrees Celsius (°C)</li> <li>identify the part played by evaporation and condensation in the water cycle and associate the rate of evaporation with temperature</li> </ul>	<p>Solid, liquid, gas, melt, boil, freeze, solidify, evaporate, condensate, degrees Celsius, state, water cycle, precipitation, transpiration</p>	<p>2. 3. 4. 5. 8. 9. <i>e.g. sort a series of everyday objects into solids, liquids and gases, based on their properties</i></p> <p>2. 3. 6. 7. 8. 9. <i>e.g. to change the temperature of a series of objects (candle wax, chocolate, water) to investigate how they change state, measuring the temperature at which they change</i></p> <p>2. 3. 5. 9. <i>e.g. set up a mini water cycle to identify the role condensation and evaporation play in the reversible process.</i></p> 



UKS2 Working Scientifically:	
1.	planning different types of scientific enquiries to answer questions, including recognising and controlling variables where necessary
2.	taking measurements, using a range of scientific equipment, with increasing accuracy and precision, taking repeat readings when appropriate
3.	recording data and results of increasing complexity using scientific diagrams and labels, classification keys, tables, scatter graphs, bar and line graphs
4.	using test results to make predictions to set up further comparative and fair tests
5.	reporting and presenting findings from enquiries, including conclusions, causal relationships and explanations of and degree of trust in results, in oral and written forms such as displays and other presentations
6.	identifying scientific evidence that has been used to support or refute ideas or arguments.

Year 5	Knowledge	Vocabulary	Skills
Autumn term	<p><b><u>Properties and Changes of Materials</u></b></p> <ul style="list-style-type: none"> <li>compare and group together everyday materials on the basis of their properties, including their hardness, solubility, transparency, conductivity (electrical and thermal), and response to magnets</li> <li>know that some materials will dissolve in liquid to form a solution, and describe how to recover a substance from a solution</li> <li>use knowledge of solids, liquids and gases to decide how mixtures might be separated, including through filtering, sieving and evaporating</li> <li>give reasons, based on evidence from comparative and fair tests, for the particular uses of everyday materials, including metals, wood and plastic</li> <li>demonstrate that dissolving, mixing and changes of state are reversible changes</li> <li>explain that some changes result in the formation of new materials, and that this kind of change is not usually reversible, including changes associated with burning and the action of acid on bicarbonate of soda</li> </ul>	<p>Solution, solute, solubility, dissolve, filter, separate, mixture, sieve, hardness, transparency, electrical conductivity, thermal conductivity, reversible change, irreversible change, permeable</p>	<p>1. 2. 3. 4. 5. 6.</p> <p><i>e.g. Investigate the effects of reversible and irreversible changes in a variety of materials, recording results</i></p> <p><i>e.g. 'Which is the best filter?' Suggest different filters to try and separate material, for example: washed gravel, washed sand, cotton wool, steel wool, muslin or nylon tights.</i></p> <p><i>e.g. classify a range of materials using an increasingly advanced techniques including testing solubility, hardness, conductivity and magnetism</i></p>



## Glossary

**Aim** – what you're trying to find out in an experiment

**Bend** – to force an object to change its shape

**Boil** – the point where a liquid reaches a temperature that it turns into a gas

**Conclusion** – a simple sentence that sums up what you found in an investigation  
**Condensate** – when a gas cools and turns into a liquid

**Degrees Celsius** – a measure of temperature(°C)

**Dissolve** – when a solid break up completely into a liquid to make a solution

**Electrical conductivity** – the measure of a material's ability to allow electricity to pass through it

**Evaporate** – when a liquid is heated up and turns into a gas

**Filter** – separating bits of solid from a liquid

**Freeze** – when a liquid is cooled down and it turns into a solid

**Gas** – one of the states of matter – very light and spreads out

**Hardness** – the quality or condition of being hard

**Irreversible change** – a change to a material's state that does not allow a return to its former state

**Liquid** – one of the states of matter – runny and takes the shape of its container

**Materials** – what something is made of

**Melt** – when a solid is heated up and turns into a liquid

**Metal** - a material which is typically hard and shiny and solid at room temperature

**Mixture** – a combination of two (or more) chemically unconnected substances

**Objects** – what something is e.g. a pencil or ruler

**Permeable** - a material which allows water to soak through it

**Plastic** – a man-made material made from organic polymers

**Precipitation** – rain, snow, sleet or hail that falls from the sky

**Prediction** – what you think will happen in an investigation

**Reversible change** – a change in a material's state that allows it to return to its former state

**Separate** – to split up a mixture of different materials

**Shiny** – a material that can reflect light

**Sieve** – a piece of equipment with small holes that allows material to be separated

**Solid** – one of the states of matter – keeps its shape and you can hold it

**Solidify** – make or become hard or solid

**Solubility** – the measure of whether something will dissolve in water

**Solution** – a mixture formed when a solid dissolves in a liquid

**State** – the form a material can take (solid, liquid or gas)

**Stretch** – to be made longer or wider without breaking

**Thermal conductivity** - the measure of a material's ability to allow electricity to pass through it

**Transparency** – see-through

**Transpiration** – water vapour passing from a plant into the atmosphere

**Water cycle** – the continuous process that allows water to circulate between the Earth's oceans and atmosphere

**Waterproof** – does not allow water to pass through it

**Wood** - the hard fibrous material that forms the main substance of the trunk or branches of a tree

**Variable** – a factor of an investigation that you will change or measure